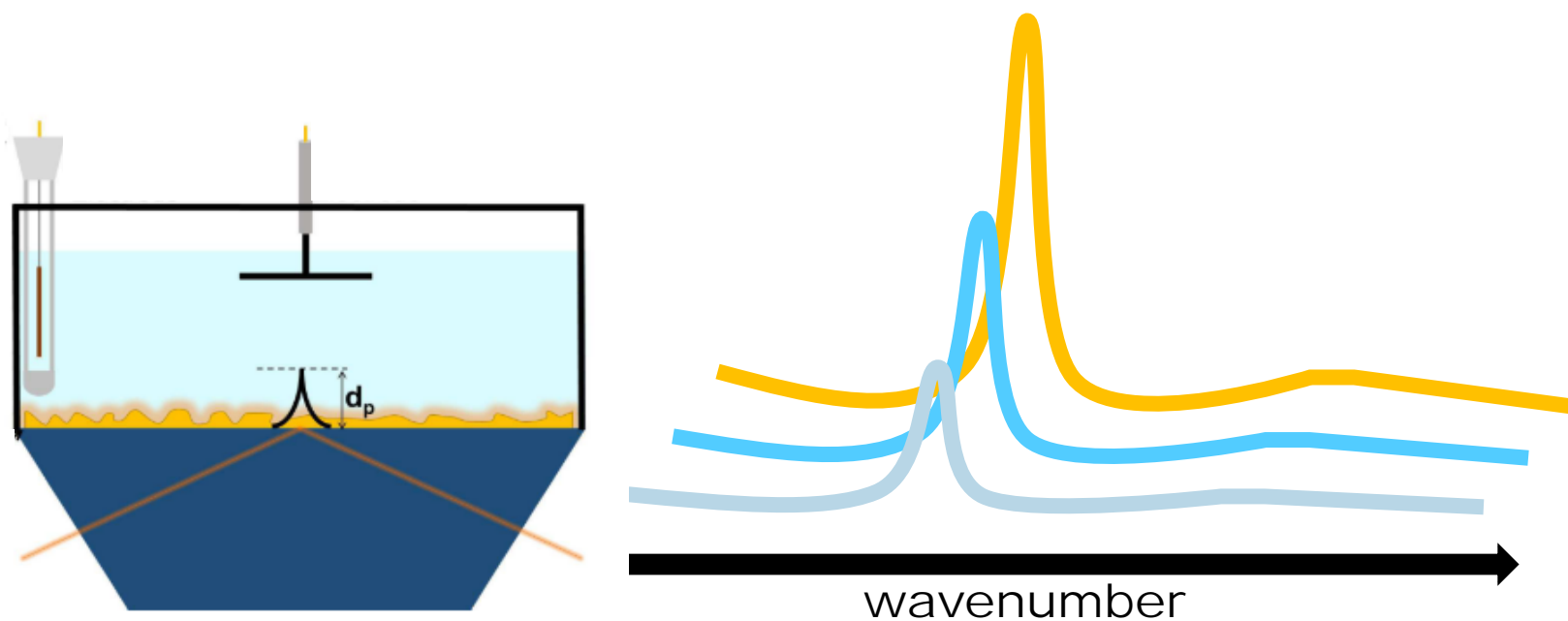


Where Electricity and Light Meet: The Power of Infrared Spectroelectrochemical Techniques in Observing Surface Chemistry



NACK webinar, April 21st

Asbury Group, Yujia Zhai (Jerry), 4th year graduate student



PennState

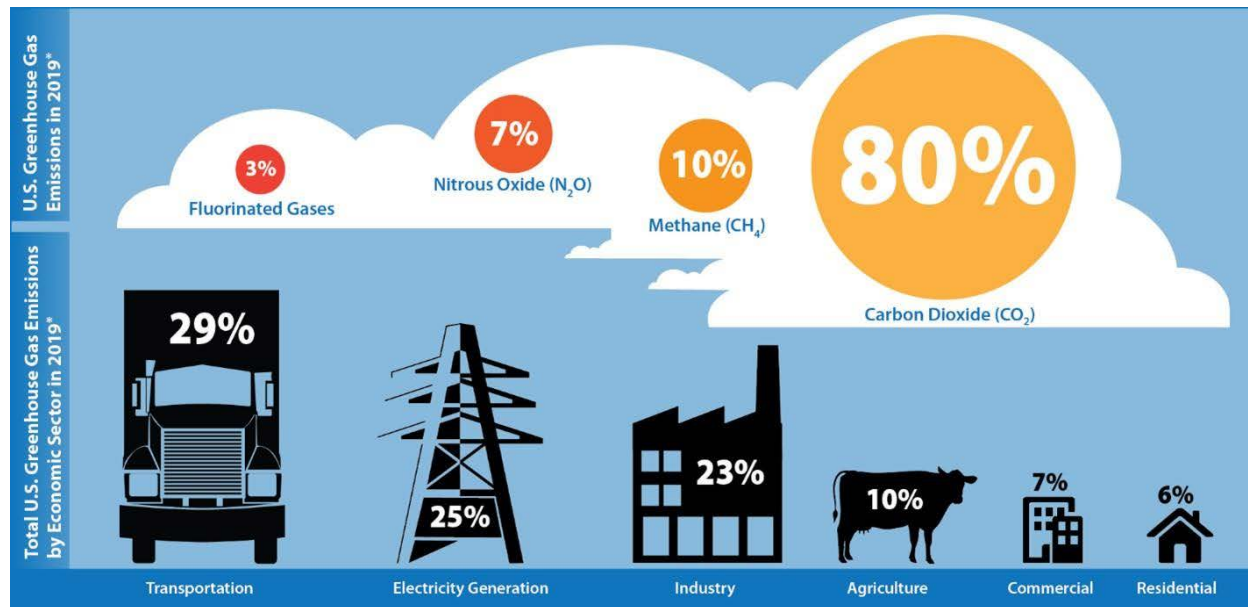
In 100 years, humans will have little reserves of fossil fuel

Fossil Fuel Resources

Fossil Fuel	World Reserves	Annual World Consumption	Depletion Time
Coal	980 billion short tons	8.3 billion short tons	118 yrs
Nat. Gas	6700 trillion cubic feet	116 trillion cubic feet	58 yrs
Oil	1480 billion barrels	32 billion barrels	46 yrs

Fossil fuel data from 2011 US EIA website (environmental information administration)

Greenhouse gas emissions cause global environmental issues.



Renewable and clean energy will replace fossil fuels.



Wind energy

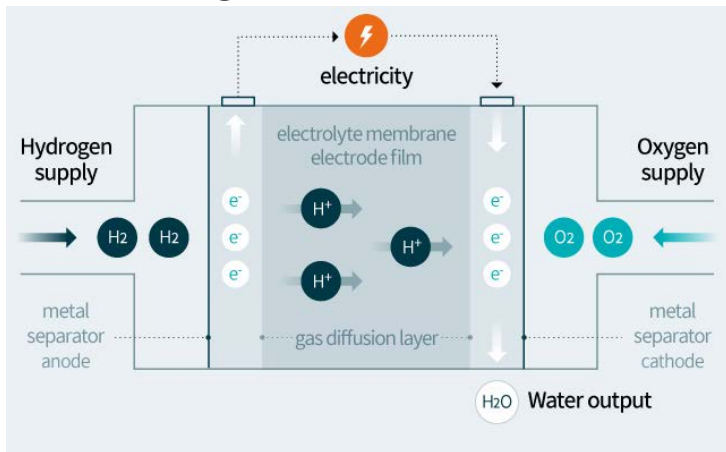


Solar energy



Hydropower energy

Electrocatalysis is promising for its low cost and wide range of usage.



Mobility



Buildings

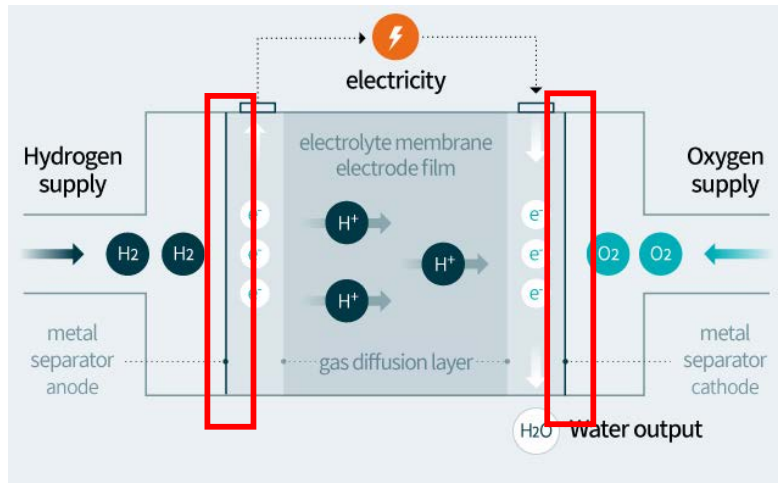


Power plant

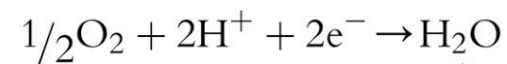


Electrochemistry is a branch of chemistry marked by the movement of electrons from one to other reactants.

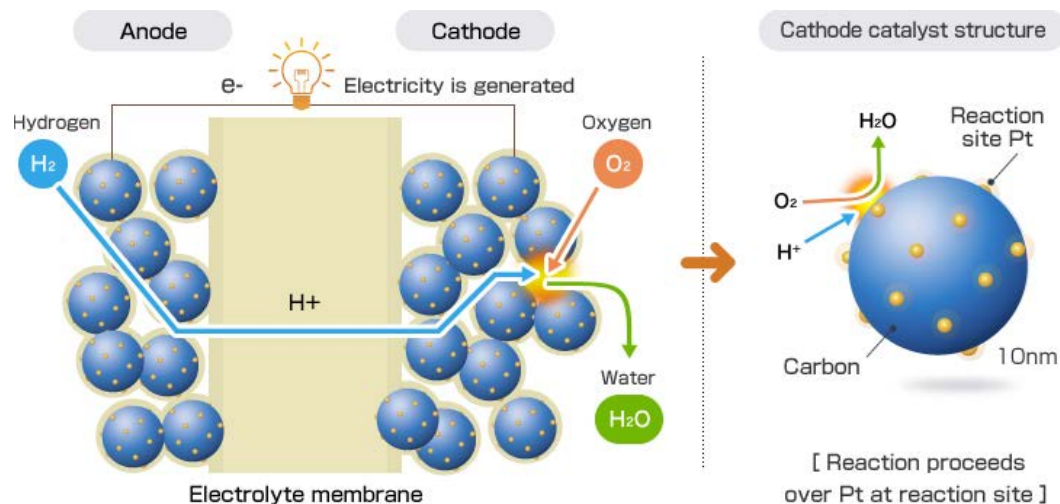
Anode reaction:



Cathode reaction:

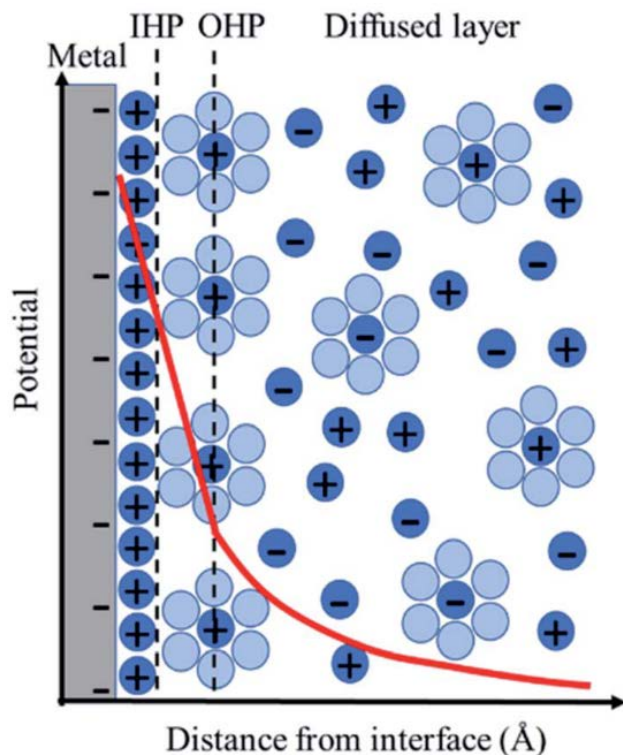


Electrocatalysis improves both reaction rate and selectivity.

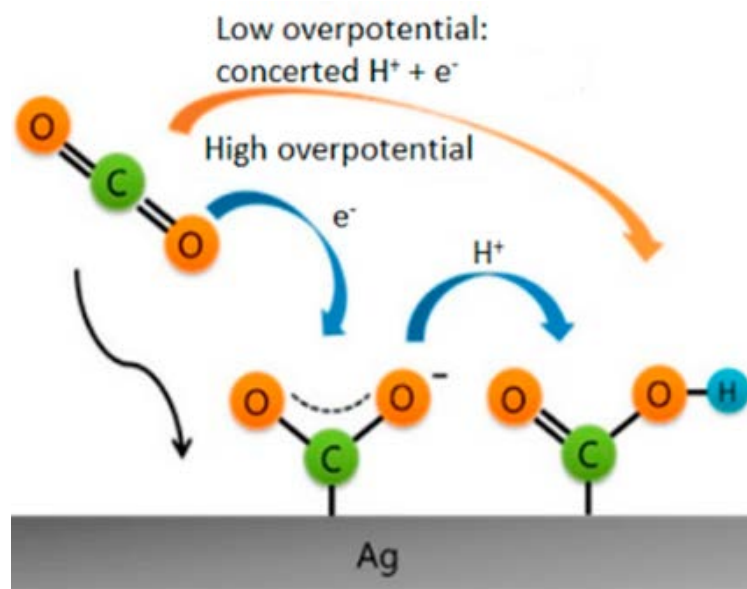


Electrocatalysis/electrochemistry is a complex surface chemistry problem.

Electrical Double layer



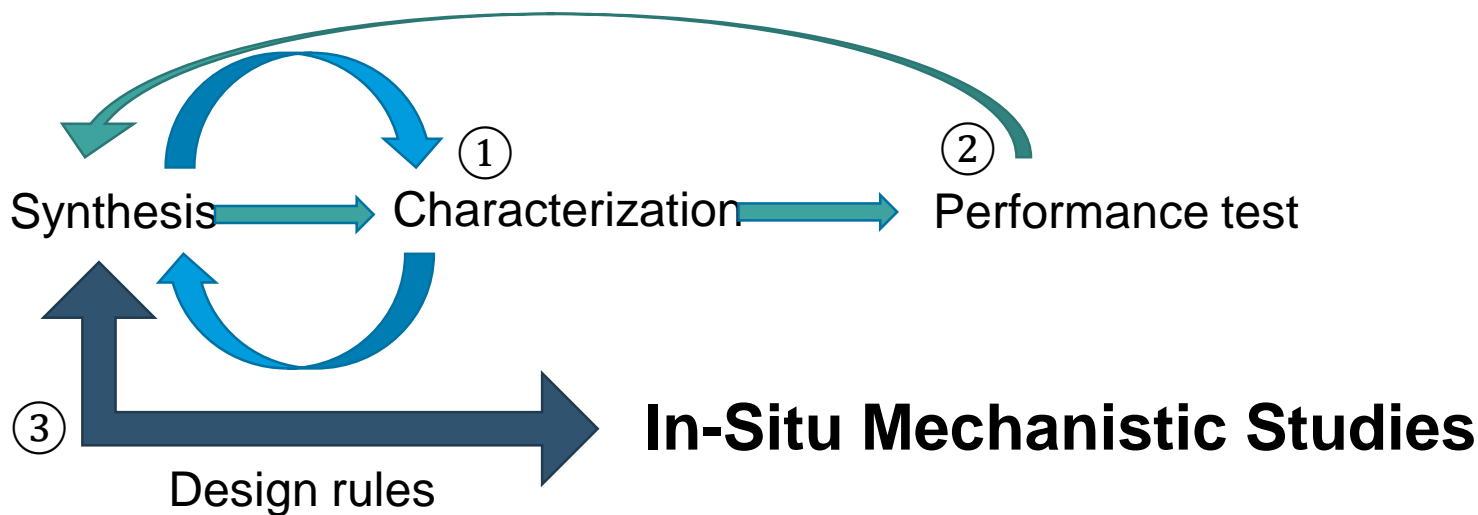
Adsorption of reactive species and intermediate



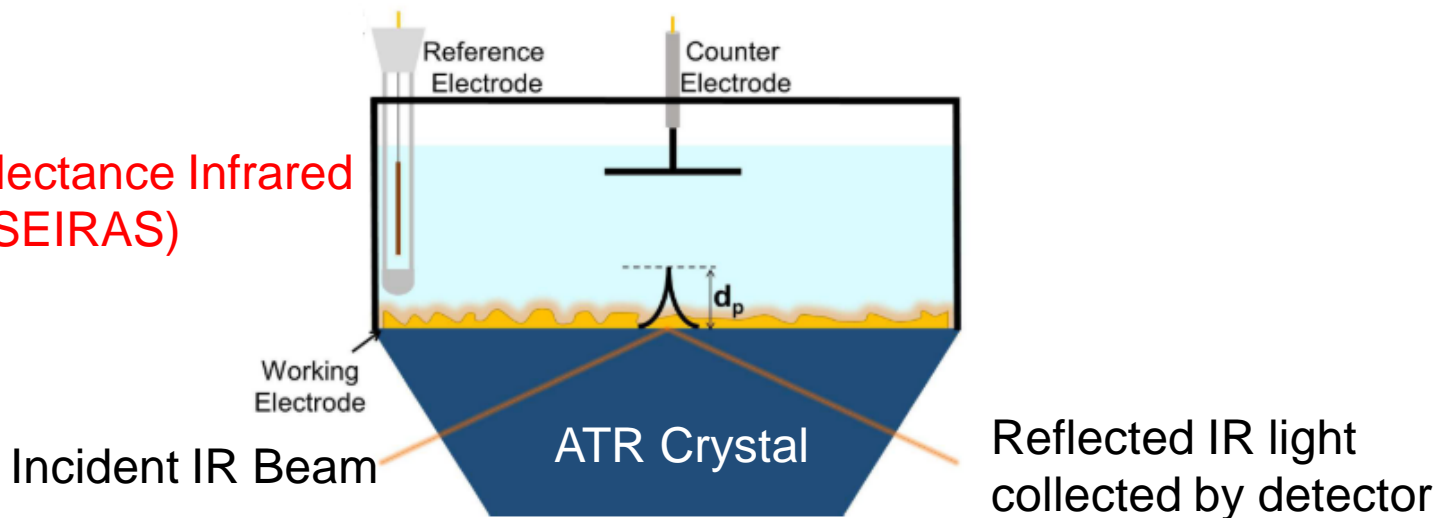
Scaling up electrocatalytic processes for industrial processes and electricity generation is still challenging.



Challenges to improve selectivity and activity of electrocatalyst remain in the field.



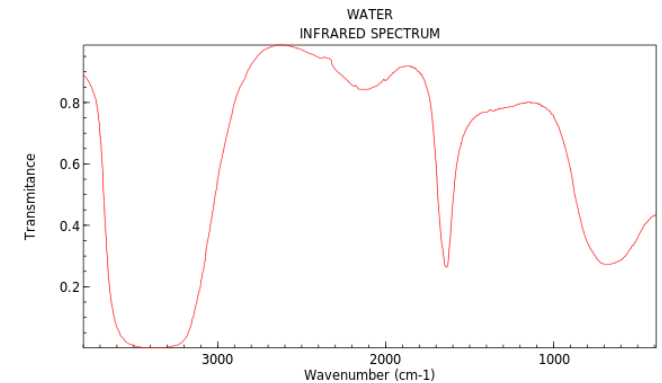
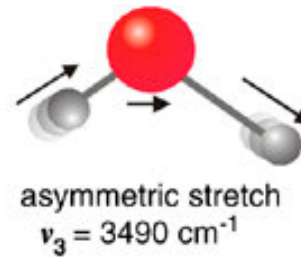
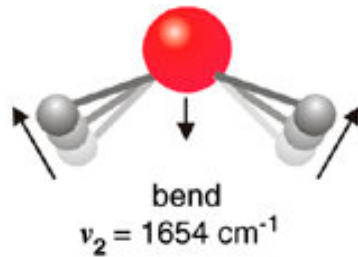
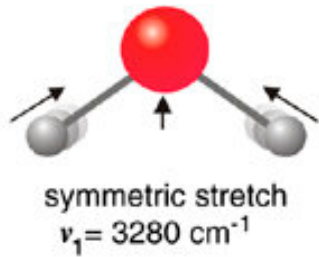
Attenuated Total Reflectance Infrared Spectroscopy (ATR-SEIRAS)



Why do we use Infrared Spectroscopy?



Molecular vibrations at certain frequencies happen upon interaction with infrared light.



Beer-Lambert's law tell us information about concentration of sample

$$A = \epsilon cl$$

A	Absorbance	
ϵ	Molar absorption coefficient	$\text{M}^{-1}\text{cm}^{-1}$
c	Molar concentration	M
l	optical path length	cm

Wavenumber/frequency of molecular vibration is related to the mass and spring constant of the bond

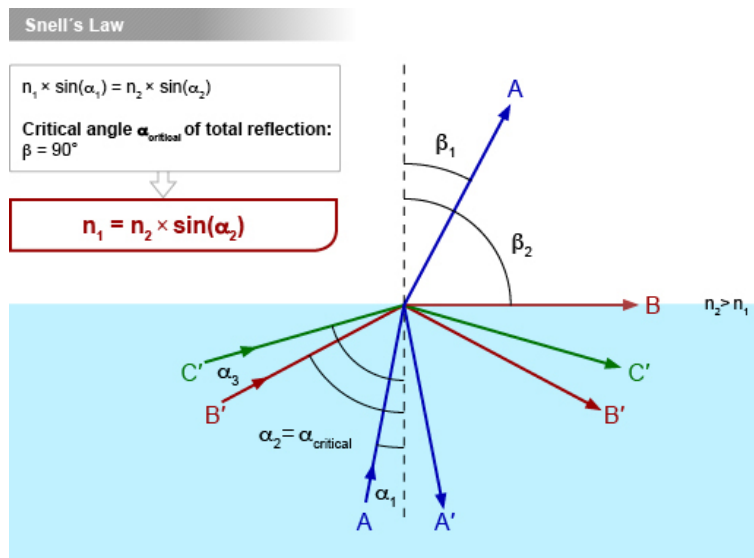
Hooke's Law

$$\nu = \frac{1}{2\pi c} \sqrt{\frac{f}{\mu}}$$

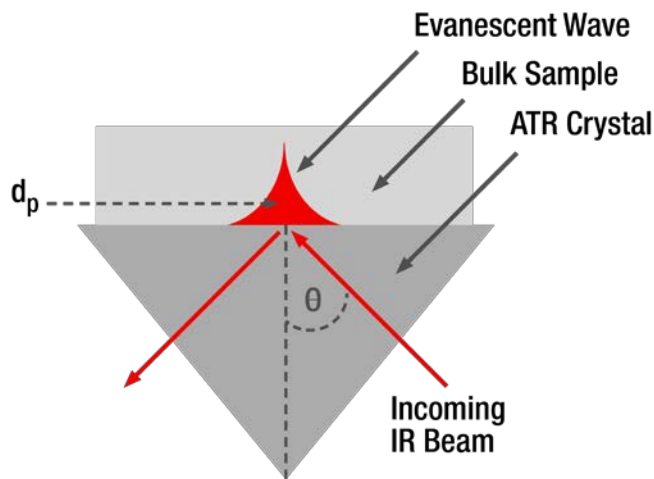
ν = frequency
 c = speed of light
 f = force constant (bond strength)
 μ = reduced mass = $\frac{m_1 m_2}{m_1 + m_2}$

ATR geometry of IR measurements

Total internal reflection and Evanescent Waves



ATR crystals	Refractive Index	α_{critical} when $n_1=1.5$
Zinc selenide (ZnSe)	2.40	40°
Germanium (Ge)	4.00	22°
Silicon (Si)	3.41	26°
Diamond	2.41	40°



$$d_p = \frac{\lambda}{2\pi n_1 \sqrt{\sin^2 \theta - \left(\frac{n_1}{n_2}\right)^2}}$$

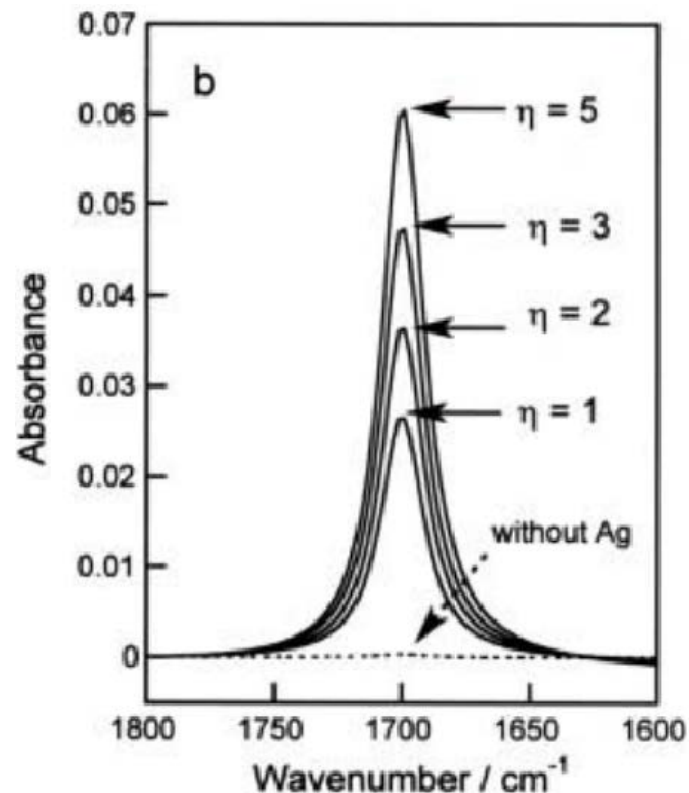
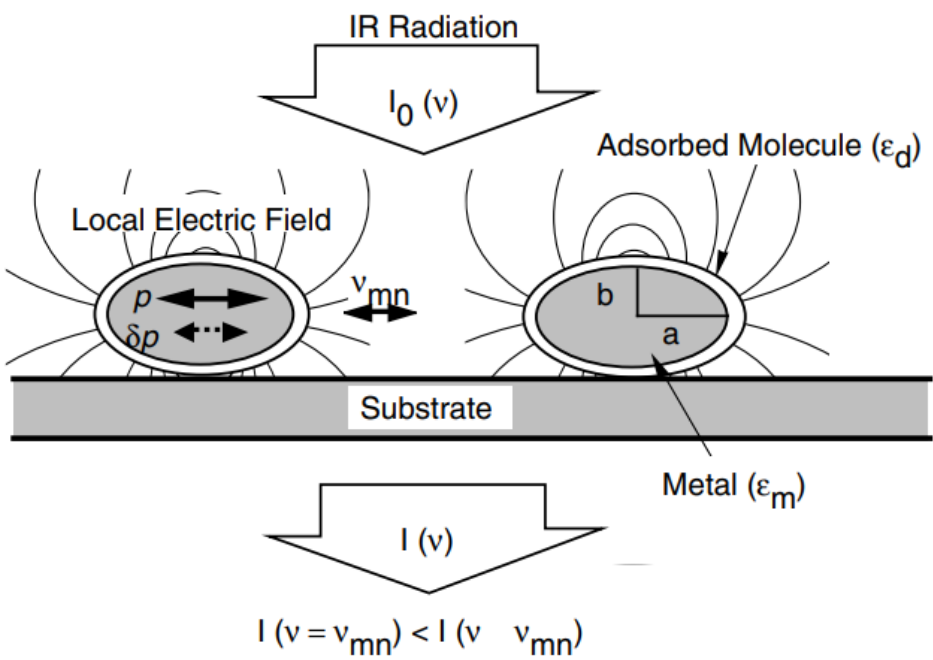
λ ... wavelength of incident light in vacuum
 n_1 ... refractive index of ATR crystal (dense medium)
 n_2 ... refractive index of sample (rare medium)
 θ ... angle of incidence



Surface enhancement principles

Localized Surface Plasmon Resonance
(Au, Ag, Cu, etc...)

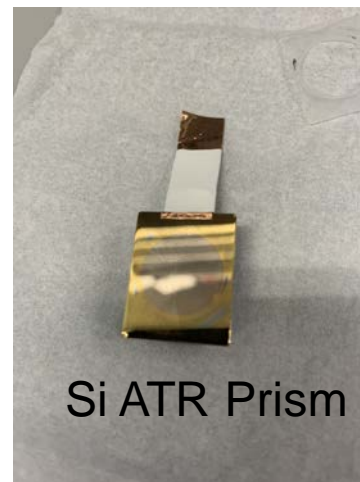
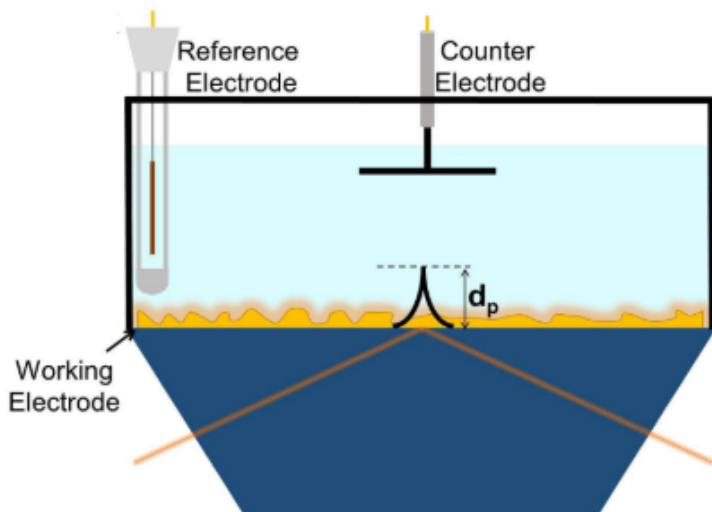
SEIRAS achieves 10–100 × enhancement of vibrational bands.



η represents the aspect ratio of the metal ellipsoid (lightning rod effect)



To achieve surface enhancement, electrode underlayer need to be prepared



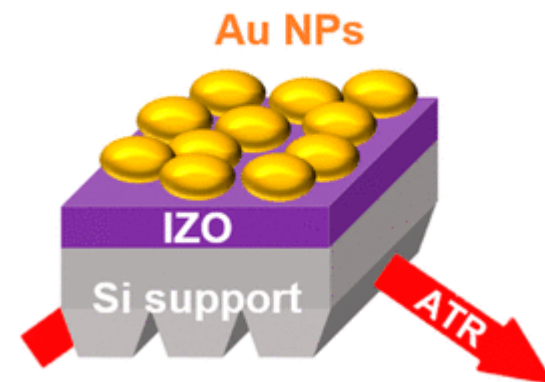
Magnetron Sputtering



Electroless plating



Electrodeposition

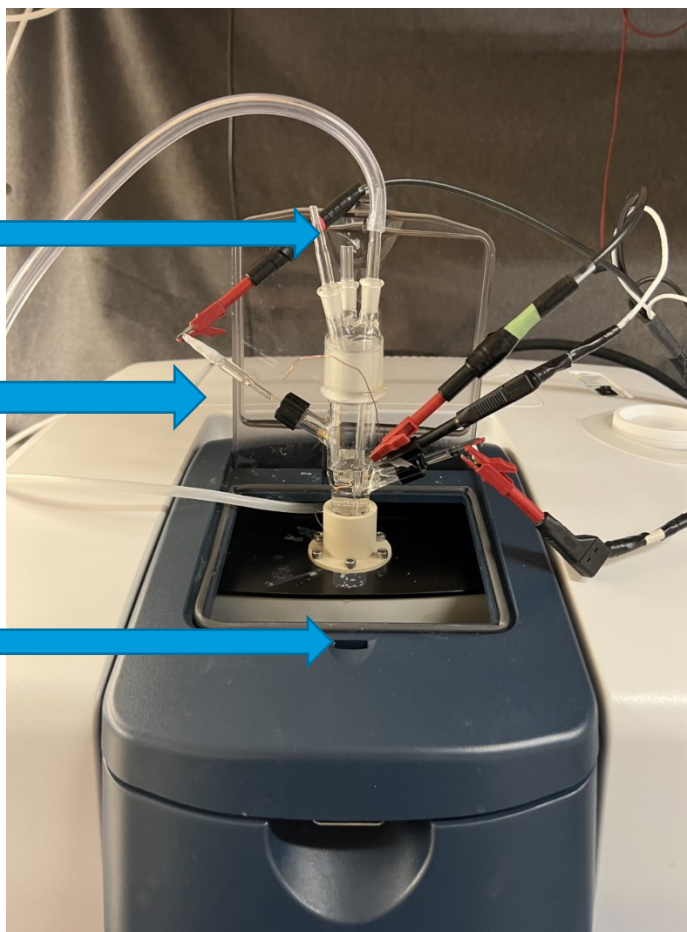


The setup of ATR-SEIRAS = ATR geometry + conductive metal film + **reaction cell**

① Gas Inlet and Outlet

② Reaction Cell
(PEEK+ Glass parts)

③ Veemax ATR Accessory



④ Counter electrode

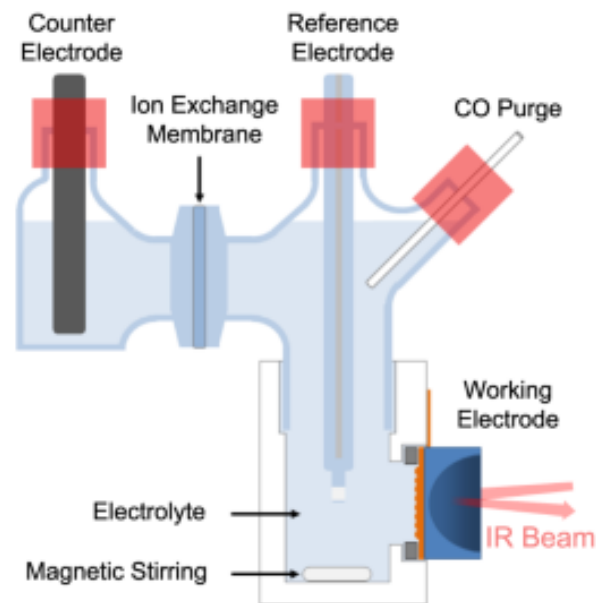
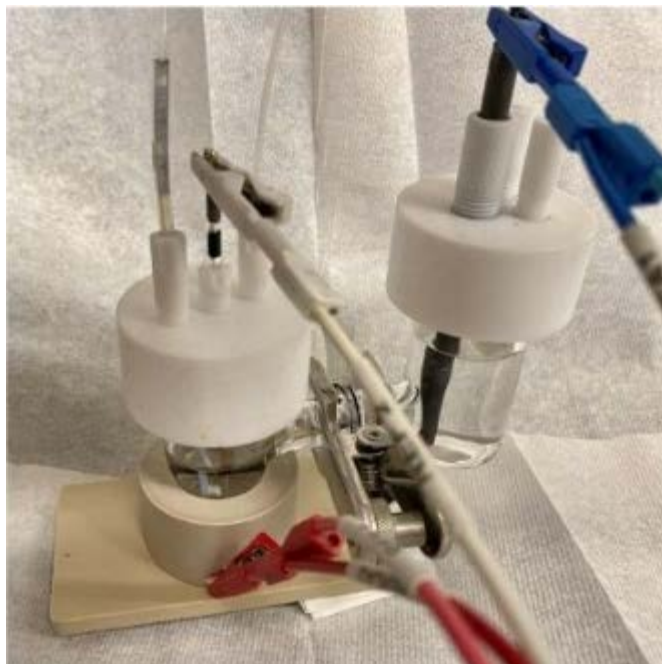
⑤ Reference
Electrode

⑥ Working Electrode



Reaction cell design can be customized for different purposes.

H-cell design Cell: Prevent Cross Contamination Stirring Cell: Improve Mass Transport



J. Am. Chem. Soc. 2022, 144, 8641–8648

Nature Communications volume 13, Article number: 2656 (2022)

Procedures for a typical ATR-SEIRAS measurement:

Sample
Preparation

Assembly of the ATR
crystal in a cell

Addition of electrolyte
and purging gas

Operation of potentiostat
and FTIR



After a series of experiment, we will gather data like below:

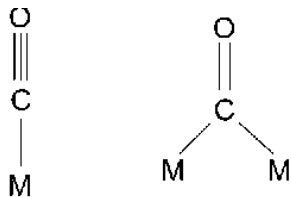
Collecting background spectrum

Background spectrum taken at 0.9 V vs RHE

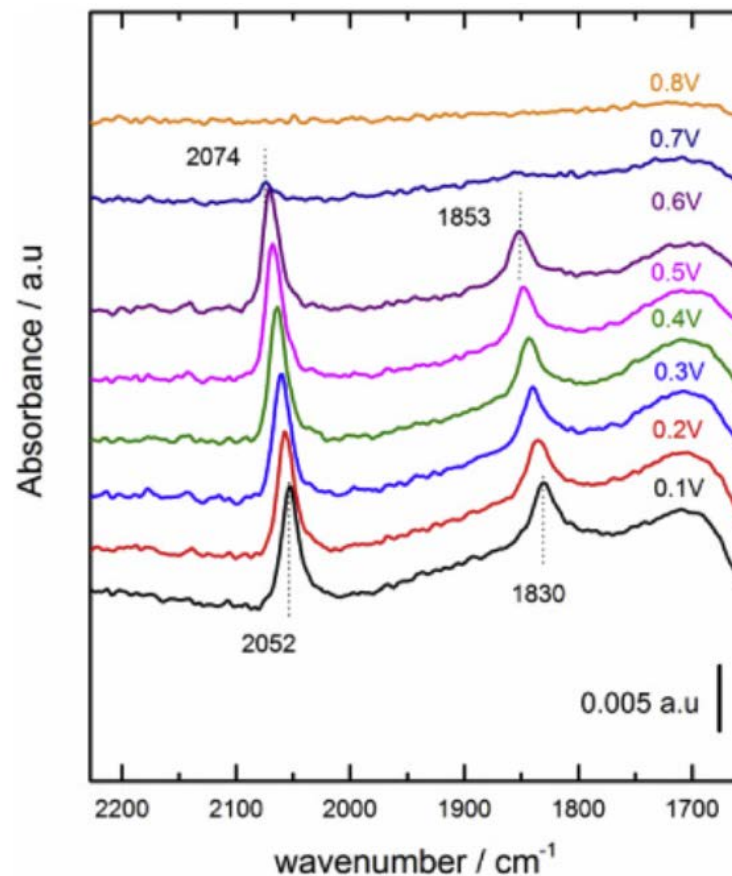
Changing applied potential and take a new spectrum

CO adsorption on Pt at different potentials

Subtracting the old spectrum from the new spectrum



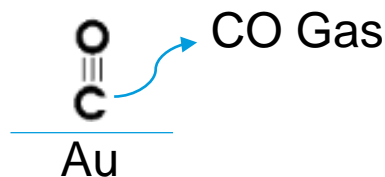
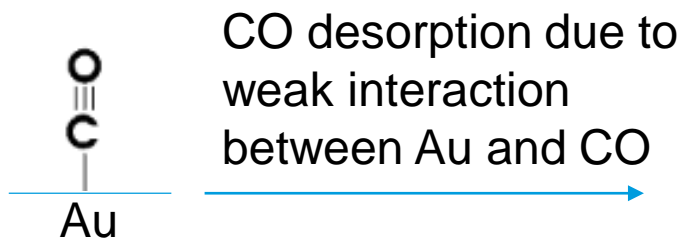
Linear-bonded Bridge-bonded



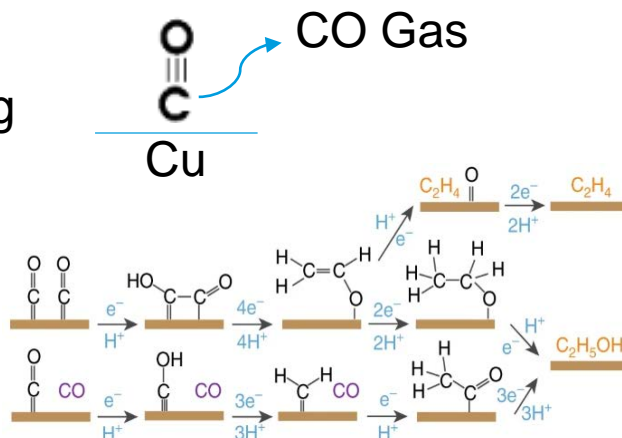
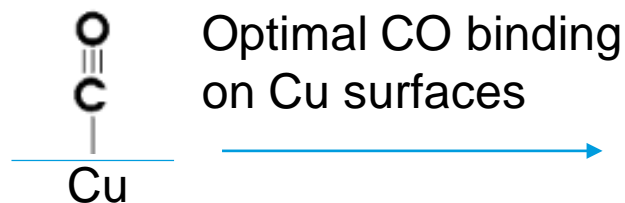
Electrochim. Acta **2018**, 281, 127– 132,



On different metals, different mechanisms were observed for the same reaction conditions.



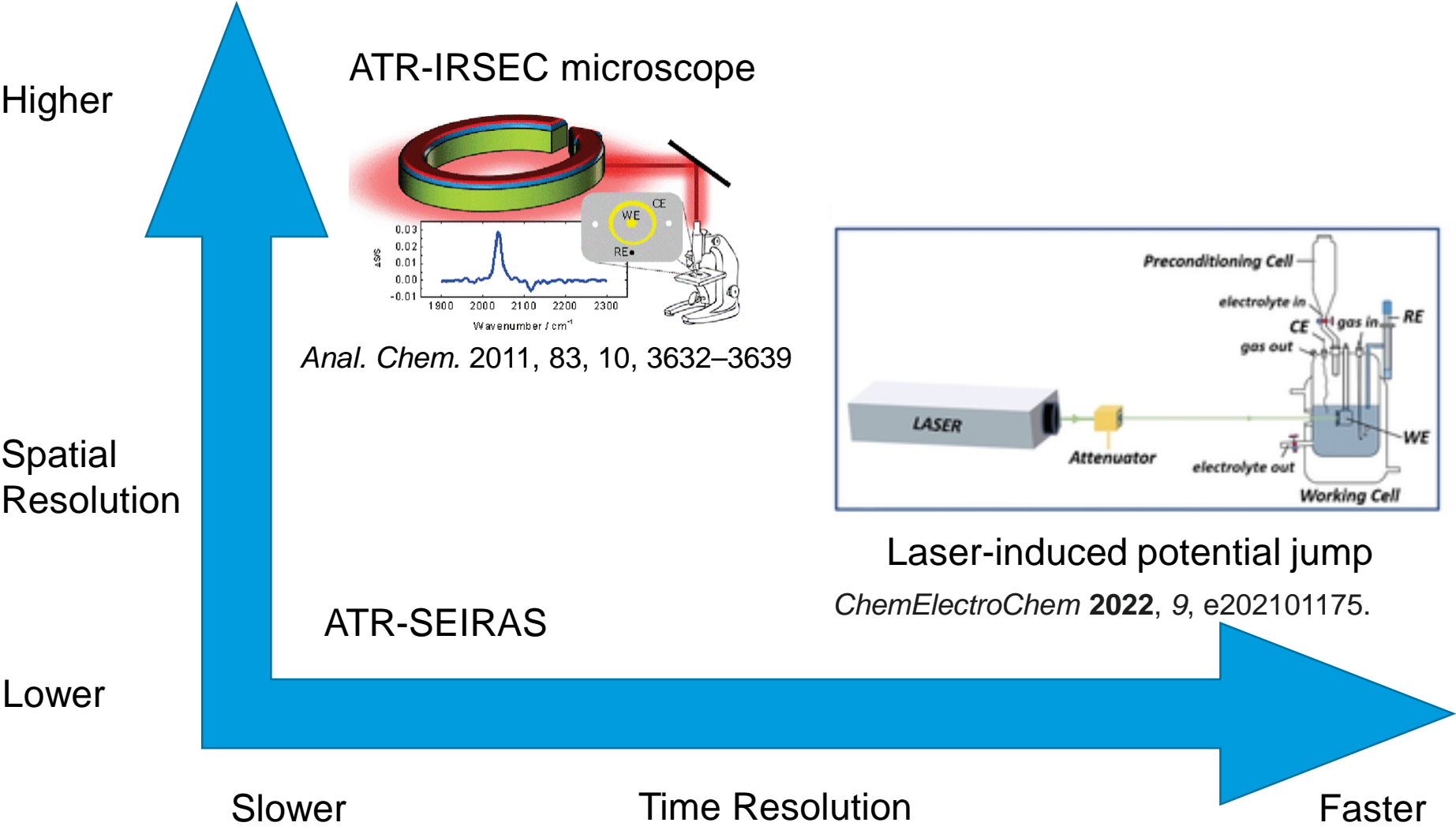
Linear-bonded CO stretch observed near 2100 cm^{-1} , no other peaks observed in the literature



Linear-bonded CO stretch observed $2070\text{-}2040\text{ cm}^{-1}$
 Multicarbon product intermediate peak reported using ATR-SEIRAS and Gas Chromatography



ATR-SEIRAS is powerful, but still limited to ensemble studies for long timescales.



Tips for starting ATR-SEIRAS:

1. Know the system and theories
2. Build/Make the cell and test its performance using samples
3. Prepare samples and prove its integrity throughout the experiment
4. Collect data under controlled reaction conditions
5. Pair With Gas Chromatography and Mass Spectrometry to detect volatile and dilute products
6. Be patient



Questions?

Join us for the next webinar!

Recent Advances in Surface Emitting Lasers



When: May 19, 2023 at 1 pm ET

Speaker: Amirhossein Ghods, staff research scientist
R&D division of ams-OSRAM

For more information, visit:
www.cneu.psu.edu/webinars



Upcoming Professional Development Opportunities

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